

# Batteries for Energy Storage

by

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## Batteries for energy storage

- Battery types
- Battery selection criteria
- Secondary batteries
- Applications
- Conclusions

## Battery Types

- Primary
- Secondary (Rechargeable)
- Reserve
- Flow batteries
- Metal air
- Fuel Cells

## Secondary batteries

- Lead acid battery
- Nickel – cadmium (NiCad)
- Nickel – metal hydride (NiMH)
- Sodium sulphur
- Sodium nickel chloride (Zebra)
- Lithium batteries

## Battery selection criteria

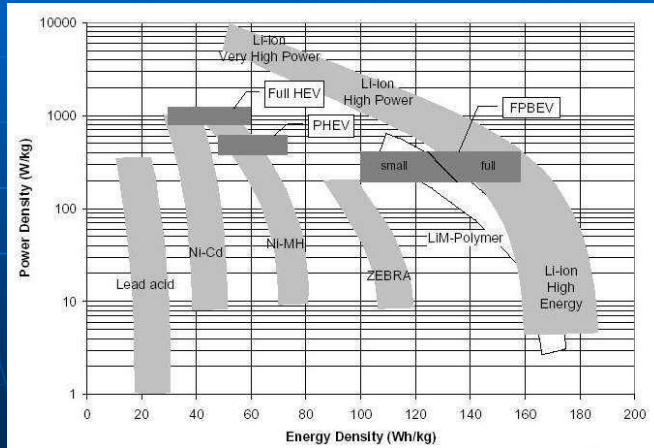
- Energy density (Wh/kg, Wh/l)
- Power density (W/kg)
- Voltage
- Power requirements and duration
- Cycle life
- Temperature (and management)
- Self discharge
- Maintenance
- Safety
- Cost
- Physical dimensions
- Charger characteristics and requirements
- Environmental (& recycling)
- Other (reliability, ease of manufacture, warranty)

## Battery selection criteria

Cell reactions & specific energies					
Battery		Cell reaction			Specific energy
+ve	-ve	Charge	Volts	Discharge	(Wh/kg)
PbO <sub>2</sub>	Pb	PbO <sub>2</sub> +2H <sub>2</sub> SO <sub>4</sub> +Pb	2.0 ↔	2PbSO <sub>4</sub> +2H <sub>2</sub> O	171 (35)
NiOOH	Cd	2NiOOH+2H <sub>2</sub> O+Cd	1.2 ↔	2Ni(OH) <sub>2</sub> +Cd(OH) <sub>2</sub>	217 (50)
NiOOH	MH	2NiOOH+MH	1.2 ↔	2Ni(OH) <sub>2</sub> +M	+280 (70)
S	Na	2Na+3S	2.0 ↔	Na <sub>2</sub> S <sub>3</sub>	760 (120)
NiCl <sub>2</sub>	Na	2Na+NiCl <sub>2</sub>	2.5 ↔	2NaCl+Ni	790 (120)
LiCoO <sub>2</sub>	LiC <sub>6</sub>	LiC <sub>6</sub> +CoO <sub>2</sub>	3.6 ↔	C <sub>6</sub> +LiCoO <sub>2</sub>	766 (140)
LiNiO <sub>2</sub>	LiC <sub>6</sub>	LiC <sub>6</sub> +Ni <sub>0.8</sub> Co <sub>0.2</sub> O <sub>2</sub>	3.7 ↔	C <sub>6</sub> +LiNi <sub>0.8</sub> Co <sub>0.2</sub> O <sub>2</sub>	766 (180)
Petrol			↔	CO <sub>2</sub>	12333

# Battery selection criteria

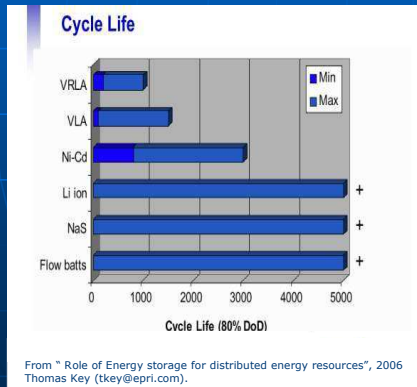
- Power and energy density



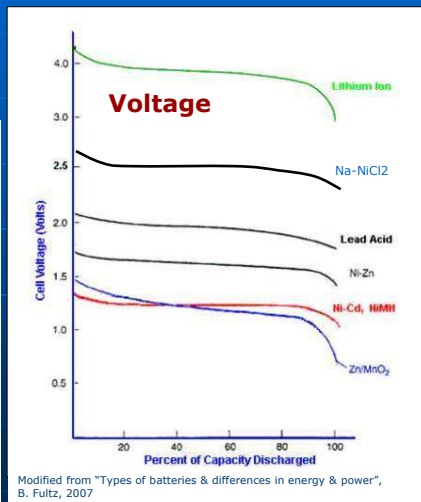
Status & prospects for zero emissions vehicle technology – Report of the ARB independent expert panel, 2007  
[http://www.arb.ca.gov/msprog/zevprog/zevreview/zev\\_panel\\_report.pdf](http://www.arb.ca.gov/msprog/zevprog/zevreview/zev_panel_report.pdf)

# Battery selection criteria

- Voltage
- Life



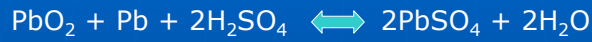
From "Role of Energy storage for distributed energy resources", 2006  
 Thomas Key (tkey@epri.com).



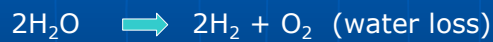
Modified from "Types of batteries & differences in energy & power",  
 B. Fultz, 2007

# Lead acid batteries

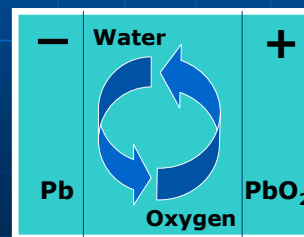
## Overall cell reaction:



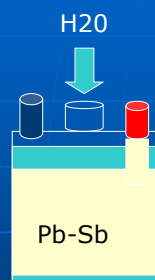
## Overcharge reaction:



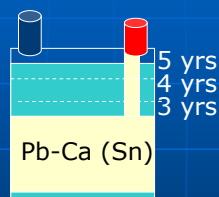
## Oxygen recombination cycle:



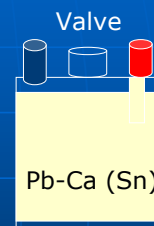
# Types of lead acid batteries



Conventional flooded or hybrid. High gas rate. Water added



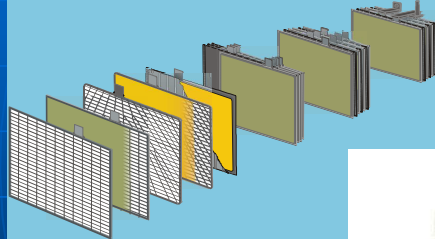
Maintenance free. "Sealed", flooded. Minor gas vented through lid. Excess water determines life



VRLA (AGM or GEL). "Sealed", no excess electrolyte. Water retained by oxygen recombination cycle. Gas may be released through valve

# SLI battery

## Construction of Cells



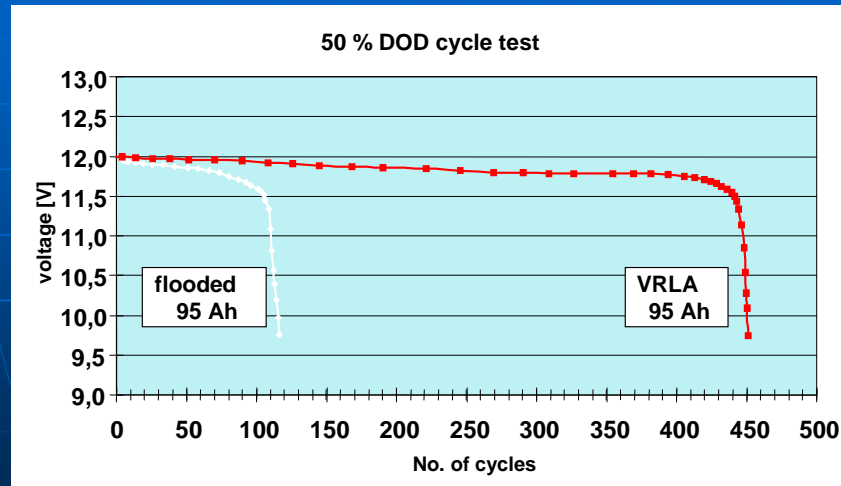
1. Maintenance free battery
2. Positive plate
3. Separator
4. Negative plate
5. Strap
6. Heat sealed lid
7. Central venting system
8. Intercell weld
9. Terminal
10. Mud rib



## SLI (Starting, lighting & ignition) batteries

- Starter motor
- Ignition
- Head Lights
- Side Lights
- Fog Lights
- Rear Lights
- Reverse Lights
- Brake Lights
- Hazard Lights
- Dashboard Lights
- Interior Lights
- Direction Indicators
- Horns
- Electronic Voice Synthesizer
- Wipers
- Screen Washer
- Fuel Pump
- Heater/Demister/Blower
- Rear Window Heater
- Rear Window Wash/Wipe
- Cigarette Lighter
- Radio/Cassette Player
- Electric Aerial.
- CB Radio
- Burglar Alarm
- Seat Belt Warning Light
- Central Locking System
- Catalyst Heater
- Electric Windows
- Electric Sun Roof
- VHF Radio Telephone
- Air Conditioning
- Electric Mirrors
- Electric Seat Adjustment
- Electric Seat Heater
- Air Damper Compressor Unit
- Water Injection System
- Electronic Fuel Injection
- Electronic Instrument Displays
- Headlamp Wash/Wipe

## SLI battery - VRLA vs flooded



## Advantages of VRLA

- Extremely low water loss
- Increased cycle life & energy throughput
- Improved cold cranking
- Increased capacity for same size
- Improved shelf life
- Leakproof battery
- No acid stratification

# Flooded lead acid batteries

## Plante battery

Thick electrodes  
20 year life  
Applications: Power stations, telecoms etc.



## Flooded tubular plate battery (OPzS)

Applications: Traction, solar and load leveling.



## Flooded flat plate battery (OGi)

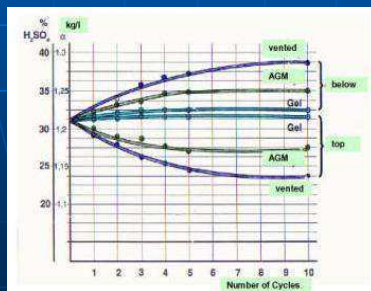
Applications: Telecoms and UPS.



<http://www.battery.co.za>

# VRLA batteries

- Valve regulated (AGM & GEL)
  - OPzV - VRLA/GEL (sealed) tubular plate
    - Applications: Telecommunications and solar
  - OGiV - VRLA/AGM (sealed) flat plate batteries
    - Applications: UPS and high rate requirements





# Sodium sulphur battery

The diagram illustrates the internal structure of a sodium sulphur battery cell. On the left, a vertical cross-section shows the following layers from top to bottom: Terminal Electrical insulation, Sodium chamber, Metal insert, Sodium electrode, Solid electrolyte, Sulfur electrode, and Cell container. On the right, a 'DISCHARGE' diagram shows the flow of ions and electrons. It labels the Negative electrode (Sodium liquid), Beta alumina (Al<sub>2</sub>O<sub>3</sub>) ceramic electrolyte, and Positive electrode (Sulphur liquid). A legend identifies the particles: Electron (small grey circle), Sodium (blue circle), Na<sup>+</sup>-Ion (green circle), Sulphur (yellow circle), and Sodiumpolysulphid (blue and yellow circles). Below this, a 3D perspective view shows a battery pack with 'Main pole' and 'Thermal enclosure' labels. To the right of the 3D view, the following specifications are listed: Cell = 2.0 V, T = 350 °C, 120 Wh/kg, 260 W/kg, Traction, UPS & load levelling. The source is cited as Illustrations: www.ngk.co.jp.

# Zebra Battery

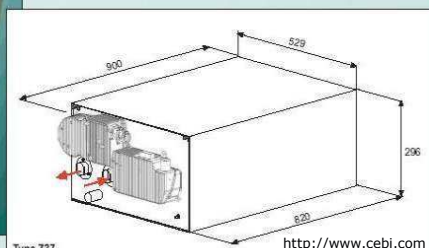
$2 \text{ NaCl} + \text{Ni} \xrightleftharpoons[\text{discharge}]{\text{charge}} \text{NiCl}_2 + 2 \text{ Na}$

**OCV 2.58 at 300°C,**  
**Operating range**  
**270°C to 350°C,**  
**Typical capacity 38Ah**  
**100% Ah-efficiency**

The diagram shows a vertical cross-section of a Zebra battery cell. It features a 'TCB Seal' at the top, a 'Current collector (+ pole)' on the right side, and a 'Cell case (- pole)' at the bottom. The internal components include 'Nickelchloride + Sodiumaluminiumchloride' and a 'Ceramic electrolyte' layer. The 'Sodium' is located at the bottom of the cell. The source is cited as <http://www.cebi.com>.

# Zebra Battery

Each battery is composed of series and parallel connected cells (example Z37 has 240 cells)  
 Battery: 120 Wh/kg, 180 W/kg (310V or 619V)  
 Cycles 3500 (80% DOD) Application: Traction



The information contained herewith is subject to change without notice

Type Z37

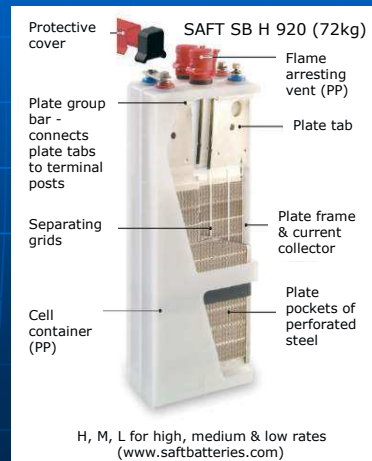
<http://www.cebi.com>

# Zebra Battery

Technical data		ZEBRA® Battery		
Type		Z37		
		Z37-310-ML3X-64	Z37-620-ML3X-32	Z37-310-ML3X-76
	id.	30x00167	30x00153	30x00268
	unit			
Capacity	Ah	64	32	76
Rated Energy	kWh	19.8	19.8	23.5
Open circuit voltage	V			
0 - 15% DOD	V	310	619	310
Max. regen. voltage	V	348	696	372
Min. op. voltage	V	206	413	206
Max. discharge current	A	224	112	224
Cell Type / N° of cells		ML3X / 240		ML3X/240
Weight with BMI	kg	201		201
Specific energy without BMI	Wh/kg	101		119
Energy density without BMI	Wh/l	154		183
Energy 2 h discharge	kWh	18		20
Specific power	W/kg	171		170
Power density	W/l	261		261
Peak power	kW	35.5 DOD 80%		33.5 DOD 70%
2/3 OCV, 30s, 335°C				
Ambient temperature	°C	-40 to +50		
Thermal loss	W	< 105		
at 270°C internal temperature				
Cooling		air		
Heating time	h	24 h at 230 VAC		
Periphery		BMI, Fan		
On board generator		HEV Application		EV Application
MAX voltage, up to 70%SOC	V/Cell	2.7		n.a.
<b>System design recommendation:</b>				
- MES-DEA Charger				
- Min. discharging time: 120 min.				
- Max. degree of discharge: 80%				
<a href="http://www.cebi.com">http://www.cebi.com</a>				

# Nickel cadmium

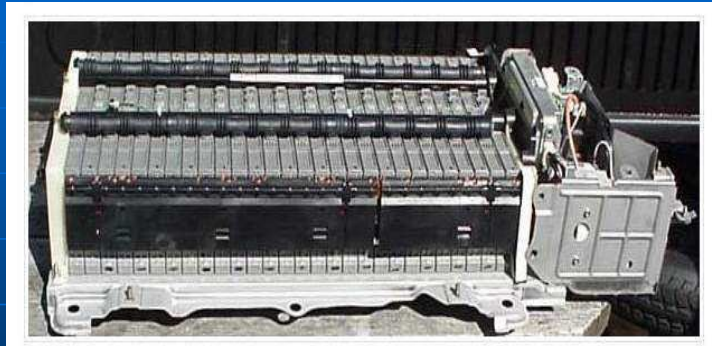
- $2\text{NiOOH} + 2\text{H}_2\text{O} + \text{Cd} = 2\text{Ni(OH)}_2 + \text{Cd(OH)}_2$  ( $E^0 = 1.29\text{V}$ )
- Temperature:  $-40\text{C}$  to  $50\text{C}$
- Electrolyte: 32% KOH
- PAM:  $\text{Ni(OH)}_2 + \text{C} + \text{Co(OH)}_2$
- NAM:  $\text{Cd(OH)}_2 + \text{Fe} + \text{C}$
- Grid: Ni plated steel
- Separators: PP, or nylon
- Cell (plate) designs:
  - **Vented** - pocket, plastic-bonded, fiber, foam or sintered
  - **Sealed** - smaller cells, cylindrical or rectangular with recombination
- Applications: UPS, telecoms, PV, HEV, EVs, satellite, aircraft, trains
- Performance: 60Wh/Kg, 200W/kg, 2000 cycles



# Nickel metal-hydride

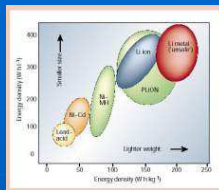
- $\text{NiOOH} + \text{MH} = 2\text{Ni(OH)}_2 + \text{M}$  ( $E^0 = 1.35\text{V}$ )
- Positive electrode like NiCd (plus ZnO &  $\text{La}_2\text{O}_3$ )
- Cell design same as sealed NiCd foam plates
- Separator: Non woven polypropylene
- Negative electrode: MH slurry on perforated foil (Ni, Cu or steel)
- Negative electrode - two types of metal hydride used:
  - $\text{AB}_5$  ( $\text{LaNi}_5$ , A=Mischmetal, La, Ce, Ti and B=Ni, Co, Mn, Al). Most used (300 Ah/kg)
  - $\text{AB}_2$  ( $\text{TiNi}_2$ , A=Ti, V and B=Ni, Zr (Cr, Co, Fe, Mn) Used by Ovonic (400Ah/kg)
- Applications: Smaller cells than NiCd. Large batteries: EV's (e.g. Toyota Prius)
- Performance: 60-70Wh/Kg, 250W/kg, 2000 cycles (80% DOD)

# Nickel metal-hydride



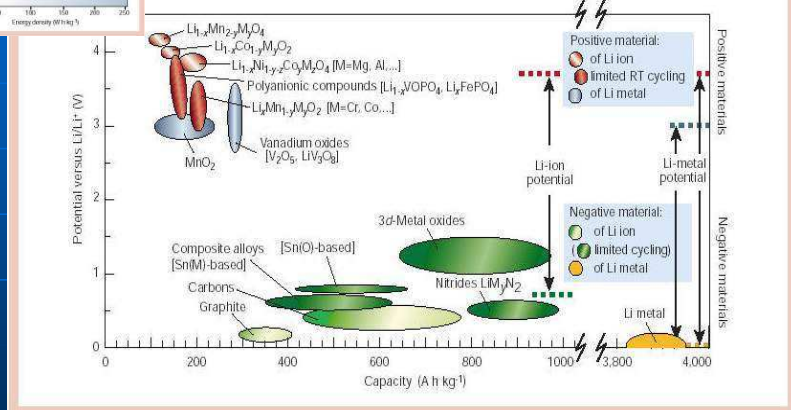
www.eaaev.org/CurrentEvents/pdf/2008/CurrentEvents200807.pdf (Photo courtesy of the Automotive Career Development Center.)

Toyota Prius (2<sup>nd</sup> generation): NiMH battery – Panasonic  
 28 modules x 6 = 168 cells, 202V, 20kW (50% DOD)  
 Air cooled & computer controlled thermal management system



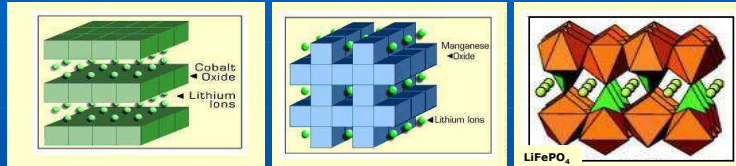
# Lithium batteries

J.M.Tarascon & M. Armand (2001), Nature, vol 414, pp359-367



Voltage versus capacity for positive and negative electrode materials for Li cells. The cell potential is represented. Note the huge difference in capacity between Li metal and other negative electrodes, hence the interest in solving the problem of Li dendrite growth.

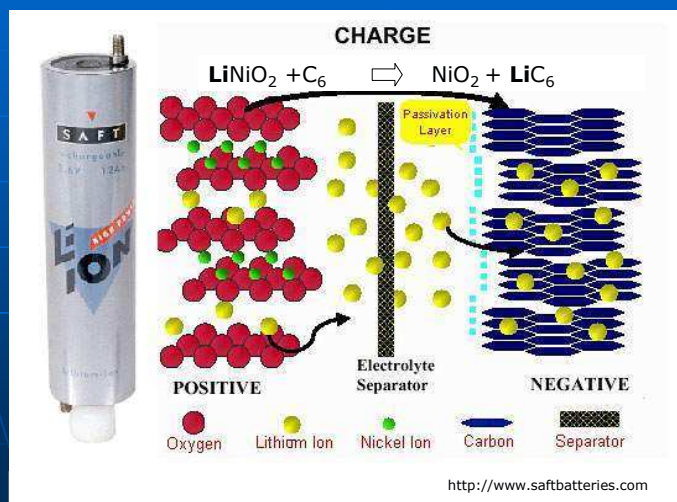
# Lithium batteries



Chemistry	Nominal Voltage (V)	Charge limit (V)	Charge & discharge C-rates	Energy density (Wh/kg)	High T stability & SoC	Active material cost (\$/kWh)	Application / comment
<b>Cobalt LiCoO<sub>2</sub></b>	3.60	4.20	1C	110-190	Fair - good	57-75	Since 1990's (Sony) Laptops.
Mn spinel LiMnO <sub>2</sub> "LMS"	3.80	4.20	10C cont. 40C pulse	110-190	Very good	25	Low energy density, high power Power tools
Li (Ni,Co,Mn) "NCM"	3.70	4.10	5C cont 30C pulse	95-130	good	30-50	Li(Ni <sub>1/3</sub> Co <sub>1/3</sub> Mn <sub>1/3</sub> )O <sub>2</sub> Power tools
Li (Ni,Co,Al) "NCA"	3.60	4.20	5C cont 30C pulse	110-190	Fair-good	50	Li(Ni <sub>0.85</sub> Co <sub>0.1</sub> Al <sub>0.05</sub> )O <sub>2</sub> Power tools
LiFePO <sub>4</sub> (A123)	3.30	3.60	35C cont	90-130	Very good	35	Lower capacity, high power

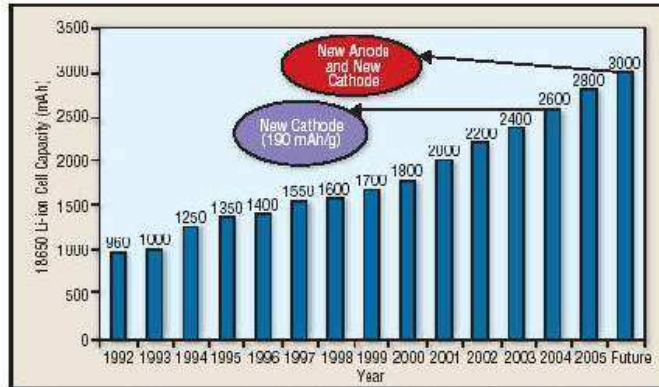
Modified from: www.BatteryUniversity.com and Wittingham, M.S. Chemical Rev, 2004 Vol 104, No 10 p4294

# Lithium batteries



<http://www.saftbatteries.com>

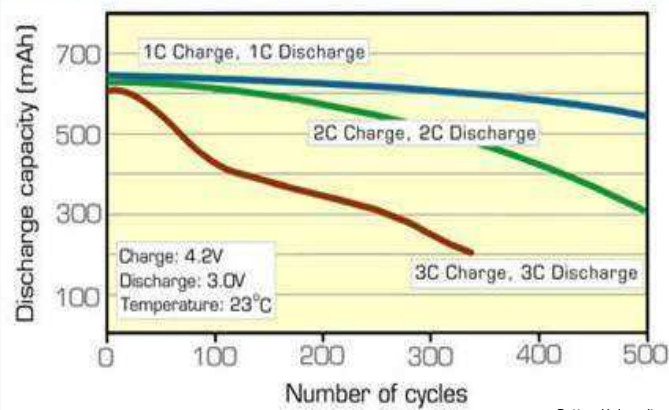
# Lithium batteries



The increase in the capacity of 18650 cells illustrates how Li-ion battery energy density has steadily risen since this battery technology was first commercialized in the early 1990s. **David Morrison**, Editor, *Power Electronics Technology*

# Lithium batteries

Cycle performance at various charge/discharge rates



www.BatteryUniversity.com

# Lithium batteries



**SAFT**  
**VL 45 E Cell**  
**3.6v, 45Ah (C/3)**  
**149 Wh/kg**  
**664 W/kg**  
**1.07 kg**

**Features**

- Very high specific energy
- Light and compact
- Maintenance free
- Excellent cycle and calendar life
- Integrated liquid cooling
- Easy integration into customized battery systems

**Applications**

- Electric and hybrid vehicles
- Telecommunication networks
- Stationary
- Space and defence



<http://www.saftbatteries.com>

**High energy lithium-ion module**  
**VLE Module**

	VLE 22-42	VLE 11-84
<b>Electrical characteristics</b>		
Nominal voltage (V)	21.6	10.8
Minimum capacity at c/3 after charge to 4.0V/cell (Ah)	42	84
Specific energy (Wh/kg)	110	111
Energy density (Wh/dm <sup>3</sup> )	158	158
<b>Specific power</b>		
(30s peak/50% DOD) (W/kg)	533	533
<b>Power density</b>		
(30s peak/50% DOD) (W/dm <sup>3</sup> )	753	753
<b>Mechanical characteristics</b>		
Height (mm)	242	242
Width (mm)	190	190
Length (mm)	123	124
Typical weight (kg)	8	8
Volume (dm <sup>3</sup> )	5.66	5.66
<b>Voltage limits</b>		
Charge (V)	4.0 (4.1 for peak/cell)	
Discharge (V)	2.7 (2.3 for peak/cell)	
<b>Current limits</b>		
Max continuous current (A)	100	300
Max 30s peak current (A)	250	500

# Lithium batteries



**SAFT - VL 34P Cell - High Power cell**

**Graphite anode, Ni alloy oxide cathode**  
**3.6v, 33Ah (C rate)**  
**2000 cycles (100 DOD)**  
**120 Wh/kg**  
**Max. Discharge: 500 A cont,**  
**1900 A 2 sec pulse**  
**Mass = 0.94 kg**  
**Diameter = 54 mm**  
**Length = 195 mm**  
**Charge CCCV TO 4.1 V (0.04)**  
**Charge current = C/2**  
**Temp D = -30 to 60 C,**  
**Temp C = 5 to 35 C**

**Applications**

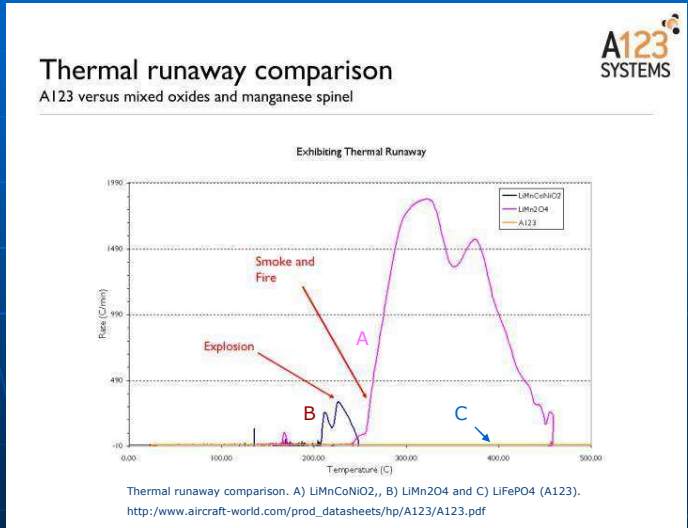
- Electric and hybrid vehicles
  - Telecommunication networks
  - Stationary
  - Space and defence
- <http://www.saftbatteries.com>

**Electrical characteristics**

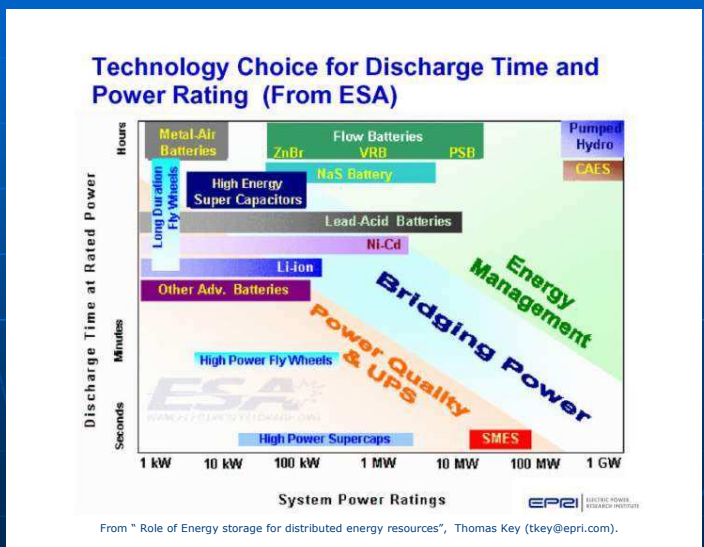
**VL 34 P module**

Nominal voltage	43.2 V
Maximum continuous current at 25° C	200 + A
Continuous power at 50% SOC	20 kW
Energy density at C rate	139 Wh/l
C rate energy	1.3 kWh
Capacity at C rate	31.5 Ah
Specific energy at C rate	75 Wh/Kg
Specific power (continuous from 100% SOC)	1100 W/Kg
Impedance at 5 sec (250 A discharge)	10 mΩ
<b>Mechanical characteristics</b>	
Length	14.6 in / 371 mm
Width	4.4 in / 112.8 mm
Height	8.3 in / 211.8 mm
Weight	35 lbs / 16 kg
<b>Operating conditions</b>	
Operating temperature range	-30° C to +60° C
Transport or storage temperature range	-40° C to +65° C
<b>Voltage limits</b>	
In charge	49.2 V
In discharge	30 V

# Lithium batteries

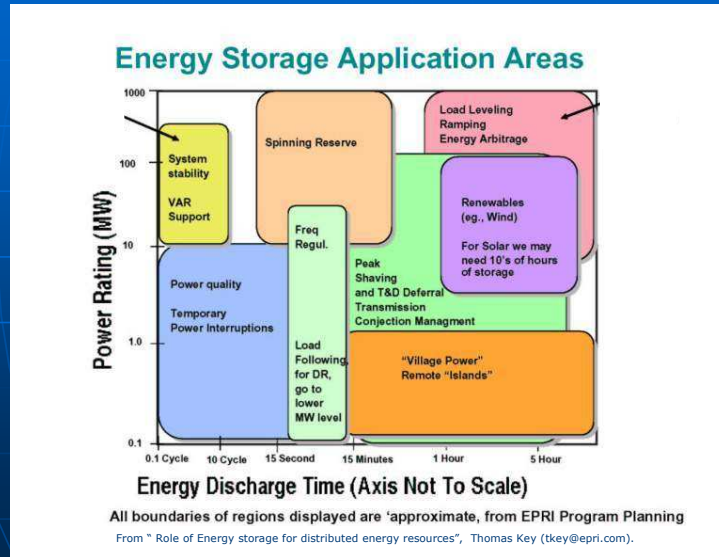


# Standby battery applications





# Battery applications



# Applications – Lead acid

**Battery systems installed between 1986 & 1997**

Technology		Power	Energy	Year	Purpose
		MW	MWh	Operation	
Lead Acid	Berlin(D)	17	14	1986 - 1995	Power Quality
					Spining reserve
Lead Acid	Tatsumi(J)	1	4	1986	Multi-purpose Demonstrator
Lead Acid	Chino(USA)	10	40	1988 - 1997	Multi-purpose
					Demonstrator
Lead Acid	South Africa	4	7	1989	Peak Shaving
					UPS
Zn-Br	Kyushu(J)	1	4	1991	Multi-purpose
					Demonstrator
Lead Acid	Puerto Rico	20	14	1994	Power Quality
					Spining reserve
Lead Acid	Vernon(USA)	3.5	3.5	1996	Peak Shaving
					UPS
Lead Acid	Metlakatla	1	1.3	1997	Grid stabilization
					Power Quality

[www.cambridgeenergy.com/archive/2005-06-22/CEF-Tarrant.pdf](http://www.cambridgeenergy.com/archive/2005-06-22/CEF-Tarrant.pdf)

# Applications - NaS & NiCad

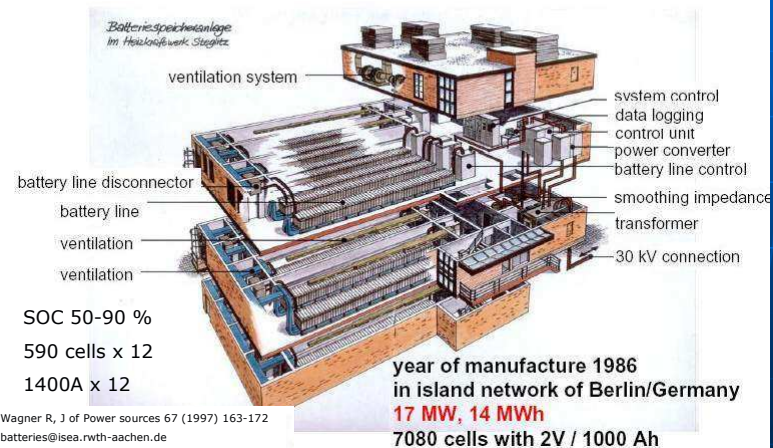
Systems Installed 1998 to 2004

Technology	Location	Power		Energy		Year	Purpose
		MW	MWh	MW	MWh		
NaS	Tsunashima Substation	6	48			1998	Load Levelling Spinning reserve
NaS	Ohito Substation	6	48			1999	Load Levelling
NaS	Saitama	2	12			1999	Load Levelling
NaS	Odaka	1	8			2000	Load Levelling
NaS	Tsunashima	2	14.4			2000	Load Levelling
NaS	Shinagawa	2	14.4			2001	Load Levelling
NaS	Kanagawa	1	7.2			2001	Peak Shaving UPS
NaS	Ebina	1	7.2			2001	Peak Shaving UPS
NaS	Chichibu Semicond pl	1	7.2			2002	load levelling UPS
Ni-Cad	Fairbanks Alaska(USA)	40	10			2003	load levelling UPS

www.cambridgeenergy.com/archive/2005-06-22/CEF-Tarrant.pdf

# Applications – Pb acid

## Battery system of BEWAG for frequency regulation



## Applications - NiCd

### Golden Valley Electric's Battery Energy Storage System (Alaska)



27 MW for 15 minutes  
13760 NiCd cells (SAFT SBH 920 cells)  
35 Mio \$  
commissioning: Aug. 2003  
batteries@isea.rwth-aachen.de

## Applications - NaS

- NaS-battery system for "load leveling" in Tokyo
- Performance characteristics
  - 2 MW
  - 1,165 V DC
  - 40 modules (12,800 cells)
  - weight > 136 tons



batteries@isea.rwth-aachen.de (NGK / TEPCO)

## Applications – Li-ion

100 kW, 15 kWh li-ion battery for UPS applications



Photo Courtesy of Saft America

## Conclusions

- There are pros and cons of every battery system which should be weighed up against the intended application
- Lead acid batteries are good for float applications, have a low initial capital cost, proven technology with high recycling efficiency and future potential for improvements in life
- Nickel cadmium and nickel metal hydride batteries have good cycling and power capabilities at increased cost. They have a memory effect and high self discharge. Metal hydride batteries have higher energy density with no toxic metals but poor charge acceptance at high temperatures and require thermal management
- Sodium metal chloride batteries have high energy density & are promising for traction and stationary applications. Disadvantages are the high temperatures, limited manufacturing facilities & lack of high power cell designs.
- Li-ion batteries are starting to enter the telecoms markets in larger cell sizes where weight & space is a criteria but at increased cost. Each cell requires electronics to control C/D voltage. Safety will remain a key issue